## ACS Review Aldehydes and Ketones - Nucleophilic Addition to the Carbonyl Group

1. What is the IUPAC name of the following compound?

A. 3-methyl-5-heptanone
B. 5-ethyl-3-hexanone
C. 5-methyl-3-heptanone
D. 2-ethyl-4-hexanone
2. Identify the correct IUPAC name of the compound below.

A. 4-benzyl-5-methylhexanal
B. 5-isopropyl-5-phenylbutanal
C. 2-methyl-3-phenylhexanal
D. 5-methyl-4-phenylhexanal
3. Which of the following is an acceptable IUPAC name for the compound below?

A. $\quad o$-bromo- $m$-chlorobenzaldehyde
B. 6-bromo-3-chlorobenzaldehyde
C. 2-bromo-5-chlorobenzaldehyde
D. 1-bromo-4-chlorobenzaldehyde
4. Which of the following has the largest $\mathrm{K}_{\mathrm{eq}}$ for the formation of the hydrate (as shown below)?

A)

C) $\mathrm{CH}_{3} \stackrel{\mathrm{O}}{\mathrm{C}} \mathrm{CHCl}_{2}$
D) $\mathrm{CH}_{3} \stackrel{\mathrm{O}}{\mathrm{C}} \mathrm{CCl}_{3}$
A. A
B. B
C. C
D. D
5. The carbon-oxygen $\pi$ bond of an aldehyde is formed by overlap of which two orbitals?
A. $\mathrm{sp}-\mathrm{sp}$
B. $\quad \mathrm{sp}^{2}-\mathrm{sp}^{2}$
C. $\quad \mathrm{sp}^{2}-2 \mathrm{p}$
D. $2 p-2 p$
6. Which of the reagents below will oxidize a secondary alcohol to a ketone?
A. $\mathrm{LiAlH}_{4}$
B. $\mathrm{HIO}_{4}$
C. $\quad \mathrm{K}_{2} \mathrm{Cr}_{2} \mathrm{O}_{7}, \mathrm{H}_{2} \mathrm{SO}_{4} / \mathrm{H}_{2} \mathrm{O}$
D. $\mathrm{HgSO}_{4}, \mathrm{H}_{2} \mathrm{SO}_{4} / \mathrm{H}_{2} \mathrm{O}$
7. Identify the reagents needed to carry out the following conversion.

A. $\quad \mathrm{H}_{2} /$ Lindlar Pd followed by $\mathrm{H}_{2} \mathrm{SO}_{4} / \mathrm{H}_{2} \mathrm{O}$
B. $\mathrm{O}_{3}$ followed by $\mathrm{H}_{2} \mathrm{O}$
C. $\mathrm{H}_{2} \mathrm{O}, \mathrm{HgSO}_{4} / \mathrm{H}_{2} \mathrm{SO}_{4}$
D. $\mathrm{LiA}_{1} \mathrm{H}_{4}$ followed by $\mathrm{H}_{2} \mathrm{O}$
8. Which one of the following works best as the reaction steps to carry out the conversion below?

A) (1) $\mathrm{Br}_{2}$
(2) $\mathrm{NaNH}_{2}(\mathrm{xs})$
(3) $\mathrm{H}_{2} \mathrm{O}, \mathrm{HgSO}_{4} / \mathrm{H}_{2} \mathrm{SO}_{4}$
B) (1) $\mathrm{B}_{2} \mathrm{H}_{6} /$ diglyme
(2) $\mathrm{H}_{2} \mathrm{O}_{2}, \mathrm{NaOH}$
(3) $\mathrm{PCC} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$
C) $(1) \mathrm{H}_{2} \mathrm{O}, \mathrm{H}_{2} \mathrm{SO}_{4}$
(2) $\mathrm{CrO}_{3} / \mathrm{H}_{2} \mathrm{SO}_{4}$
D)

(2) $\mathrm{NaOH}, \mathrm{H}_{2} \mathrm{O}$
A. A
B. B
C. C
D. D
9. Which of the following reagents would carry out the isotopic substitution reaction shown below?

A. $\quad{ }^{18} \mathrm{O}_{2} / \mathrm{Ni}$ (cat.)
B. $\mathrm{H}_{2}{ }^{18} \mathrm{O} / \mathrm{HCI}$ (cat.)
C. $\mathrm{Cr}^{18} \mathrm{O}_{3} /$ pyridine
D. ${ }^{18} \mathrm{O}_{3}$
10. What is the product of the following reaction?

A. 3-methylpentane
B. 3-methyl-2-pentanol
C. 3-methyl-2-pentene
D. 3-methyl-1-pentyne
11. What is the product of the reaction below?

A)

B)

C)

D)

A. A
B. B
C. C
D. D
12. The compound shown below is the hemiacetal formed between:

A. propanal and 2-propanol
B. 2-methylpropanal and ethanol
C. acetone and 1-propanol
D. ethanal and 2-methyl-1-propanol
13. What is the product of the reaction of butanal with excess methanol and catalytic sulfuric acid?
A) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}$
B) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{COCH}_{3}$
C) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OCH}_{3}$
D) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{OCH}_{3}\right)_{2}$
A. A
B. B
C. C
D. D
14. Identify the products of the hydrolysis of the following compound.
$\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{OCH}_{3}\right)_{2} \xrightarrow{\mathrm{H}_{2} \mathrm{O}, \mathrm{H}^{+}}$
A) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \stackrel{\mathrm{O}}{\mathrm{C}} \mathrm{H}+2 \mathrm{CH}_{3} \mathrm{OH}$
B) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}+2 \mathrm{H}_{2} \mathrm{C}=\mathrm{O}$
C) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \stackrel{\mathrm{O}}{\mathrm{COCH}_{3}}+\mathrm{CH}_{3} \mathrm{OH}$

## D) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}+2 \mathrm{CH}_{3} \mathrm{OH}$

A. A
B. B
C. C
D. D
15. Which one of the following is not an intermediate in the acid-catalyzed reaction of benzaldehyde with 2 equivalents of methanol to give benzaldehyde dimethyl acetal?

I

II

III

IV
A. I
B. II
C. III
D. IV
16. The compound shown below is the cyclic hemiacetal of:

A. 5-hydroxyheptanal
B. 5-hydroxy-2-heptanone
C. 6-hydroxy-3-heptanone
D. 6-hydroxyheptanal
17. What is the product of the reaction shown?

A)

B)

C)

D)

A. A
B. B
C. C
D. D
18. Acid-catalyzed hydrolysis of the cyclic acetal below gives:

A. ethanal and 2-chlorocyclohexanol
B. 1,2-ethanediol and 2-chlorocyclohexanol
C. ethanol and 2-chlorocyclohexanol
D. 1,2-ethanediol and 2-chlorocyclohexanone
19. What are the products of the following reaction?

A. cyclohexanone and ethanol
B. cyclohexanone and ethanal
C. 1,2-cyclohexanediol and ethanal
D. 1,2-cyclohexanediol and ethanol
20. The acid-catalyzed reaction of propanal with 2 equivalents of methanol forms an acetal. This can mechanistically be thought of as:
A. an addition reaction followed by a substitution reaction
B. a substitution reaction followed by an addition reaction
C. an elimination reaction followed by a substitution reaction
D. an addition reaction followed by an elimination reaction
21. Which one of the following gives ethanal, $\mathrm{CH}_{3} \mathrm{CH}=\mathrm{O}$, (as one of two products) when added to an aqueous solution of HCl ?
A)

B)

C)

D)

A. A
B. B
C. C
D. D
22. Which synthetic method below correctly does the following conversion?

A) (1) $\mathbf{M g}$, diethyl ether
(2) $\mathrm{D}_{2} \mathrm{O}$
B) (1) $\mathrm{LiAlD}_{4}$, diethyl ether
(2) $\mathrm{D}_{2} \mathrm{O}$
C) (1) $\mathrm{HOCH}_{2} \mathrm{CH}_{2} \mathrm{OH}, \mathrm{H}^{+}$
(2) Mg , diethyl ether
(3) $\mathrm{D}_{2} \mathrm{O}$
(4) $\mathrm{H}_{2} \mathrm{O}, \mathrm{H}^{+}$
D) (1) $\mathrm{HOCH}_{2} \mathrm{CH}_{2} \mathrm{OH}, \mathrm{H}^{+}$
(2) DCl
(3) $\mathrm{H}_{2} \mathrm{O}, \mathrm{H}^{+}$
A. A
B. B
C. C
D. D
23. What is the product of the reaction below?


A. 2-methyl-1-pentene
B. 2-methyl-2-propyloxirane
C. 4-methyl-1-pentene
D. 1-pentene
24. Domiodol, shown below, is used medicinally as a mucolytic agent. What are the acid-catalyzed hydrolysis products of Domiodol?

A) ICH $\underset{\substack{\mathrm{OH}}}{\mathrm{HOCH}_{2} \mathrm{CHCH}_{2} \mathrm{OH}}$
B) $\mathrm{ICH}_{2} \mathrm{COH}+\mathrm{HOCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}$
C) $\mathrm{ICH}_{2} \mathrm{OH}+\underset{\mathrm{HOCH}_{2} \mathrm{CHCH}}{\stackrel{\mathrm{O}}{\mathrm{CH}}}$
D) $\mathrm{ICH}_{2} \mathrm{CH}+\mathrm{HOCH}_{2} \mathrm{CH}_{2} \stackrel{\mathrm{O}}{\mathrm{CH}} \mathrm{H}$
A. A
B. B
C. C
D. D
25. Which of the following reacts with $\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right)_{2} \mathrm{NH}$ to give the compound shown below?

## $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}=\mathrm{CHN}\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)_{2}$

A) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{Br}$
B) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \stackrel{\mathrm{O}}{\mathrm{C}} \mathrm{H}$
C) $\mathrm{CH}_{3} \mathrm{CH}_{2} \stackrel{\mathrm{O}}{\mathrm{C}} \mathrm{CH}_{3}$
D) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \stackrel{\mathrm{O}}{\mathrm{O}} \mathrm{O}$
A. A
B. B
C. C
D. D
26. What is the product of the following reaction sequence?
$\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{Br} \xrightarrow[\text { (2) } \mathrm{CH}_{3} \mathrm{Li}]{\text { (1) } \mathrm{P}\left(\mathrm{C}_{6} \mathrm{H}_{5}\right)_{3}} \xrightarrow{\text { cyclopentanone }}$
A)

B)

C)

D)
$\mathrm{CHCH}_{2} \mathrm{CH}_{3}$
A. A
B. B
C. C
D. D
27. Baeyer-Villiger oxidation reactions can use peroxycarboxylic acids to convert ketones to:
A. carboxylic acids
B. esters
C. epoxides
D. $\alpha$-hydroxy ketones
28. Identify the missing reagent for the reaction below.

A)

B)

C)

D)

A. A
B. B
C. C
D. D
29. What is the product of the following Baeyer-Villiger oxidation reaction?

A)

B)

C)

D)

A. A
B. B
C. C
D. D
30. Which of the following is the product of the reaction between acetone, $\mathrm{CH}_{3} \mathrm{COCH}_{3}$, and methylamine, $\mathrm{CH}_{3} \mathrm{NH}_{2}$ ?
A) $\mathrm{CH}_{3} \mathrm{CCH}_{3}$
B) $\mathrm{CH}_{3} \stackrel{\text { - }}{\mathrm{C}}=\mathrm{CHNCH}_{3}$
C) $\mathrm{CH}_{3} \stackrel{\text { CCH }}{3}$
$\mathrm{NH}_{2}$
D)

A. A
B. B
C. C
D. D
31. Which of the following reacts with methylamine at the fastest rate?
A. 1-pentene
B. pentanal
C. 2-pentanone
D. 3-pentanone
32. Which of the following gives the greatest percentage of hydrate (gem-diol) when dissolved in water?
A. butanal
B. 2,2-dichlorobutanal
C. 3,3-dichlorobutanal
D. 4,4-dichlorobutanal
33. Which of the following methods can be used to synthesize 2-methyl-1-hexene, shown below, with no formation of isomeric by-products?

A)

B)

C)

D)


A. A
B. B
C. C
D. D
34. Which of the following is the best method to synthesize 2-methyl-3-pentene, shown below, with little or no by-product formation?

A)

B)

C)

D)


A. A
B. B
C. C
D. $D$
35. What is the product of the reaction sequence below?

A. 2-methyl-1-hexene
B. 2,3-dimethyl-2-pentene
C. 2-methyl-2-hexene
D. 3-methyl-1-hexene
36. Which of the following is a hemiacetal?
A)

B)

C)

D)

A. A
B. B
C. C
D. D

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1. C
2. D
3. C
4. D
5. D
6. C
7. C
8. B
9. в
10. A
11. A
12. A
13. D
14. A
15. в
16. в
17. A
18. D
19. С
20. A
21. C
22. C
23. A
24. A
25. в
26. D
27. в
28. A
29. в
30. A
31. в
32. в
33. D
34. C
35. D
36. C
